

# Preparation and Properties of Cardiac Cytochrome $c_1$ <sup>†</sup>

Chong H. Kim\* and Tsao E. King

Departments of Chemistry and Biology and Laboratory of Bioenergetics, State University of New York at Albany,  
Albany, New York 12222

Received August 20, 1986; Revised Manuscript Received December 9, 1986

**ABSTRACT:** A method for the large-scale isolation of beef heart mitochondrial cytochrome  $c_1$  in high purity was developed. This method gave higher yield of "one-band" cytochrome  $c_1$  than previously reported [Kim, C. H., & King, T. E. (1981) *Biochem. Biophys. Res. Commun.* 102, 607-614]. In addition, the present method was effective in the preparation of "two-band" cytochrome  $c_1$  which was used to prepare the hinge protein according to the principle of sequential resolution [Kim, C. H., & King, T. E. (1983) *J. Biol. Chem.* 258, 13543-13551]. The isolation of one-band and two-band cytochrome  $c_1$  by this procedure could be completed within 3 or 4 days starting with succinate-cytochrome  $c$  reductase. One-band cytochrome  $c_1$  showed a molecular weight of 44 000 by sedimentation equilibrium and 29 000 by sodium dodecyl sulfate-polyacrylamide gel electrophoresis. The disparities in these data from the actual value of 27 924 by amino acid sequence analysis, as previously reported [Wakabayashi, S., Matsubara, H., Kim, C. H., & King, T. E. (1982) *J. Biol. Chem.* 257, 9335-9344], are most probably due to the formation of detergent or detergent-phosphate complex. A comparison of some properties of one-band cytochrome  $c_1$  with those of two-band cytochrome  $c_1$  clearly showed significant differences between the two preparations. These results suggest the hypothesis that one of the possible roles of the hinge protein in the mitochondrial respiratory chain is to stabilize the conformation of cytochrome  $c_1$ .

Cytochrome  $c_1$  was independently discovered by Keilin and Hartree in Cambridge (Keilin, 1966) and Yakushiji and Okunuki in Japan (Yakusiji & Okunuki, 1940). However, the preparation of pure and active cytochrome  $c_1$  was not successful until King and co-workers reported a method in 1972 (King, 1978; Yu et al., 1972), and, more recently, van Gelder and co-workers reported another method in 1980 (König et al., 1980). However, the sample prepared even by the latter method was still heavily contaminated with the glutamic acid rich small protein known as the "hinge protein" [cf. Wakabayashi et al. (1982a) and Kim & King (1983)]. The evidence was the amino acid composition of this preparation. Our cytochrome  $c_1$  preparation reported in 1972 was actually cytochrome  $c_1$  with the hinge protein in a 1 to 1 molar ratio, which we called "two-band"  $c_1$ , and similarly, the cytochrome  $c_1$  preparation free of the hinge protein was called the "one-band"  $c_1$  (Kim & King, 1981, 1983).

In our preliminary report (Kim & King, 1981), the yield of the preparation of one-band cytochrome  $c_1$  free of the hinge protein was very low (<10%). During the last few years, we have substantially improved the yield of cytochrome  $c_1$  (ca. 25%). Moreover, we have found that the yield of one-band  $c_1$  was closely related to that of two-band  $c_1$ .

This paper reports an improved method in a relatively large scale of cytochrome  $c_1$  isolation. Many properties of one-band cytochrome  $c_1$  were studied, and a comparison in the behavior between the pure one-band and two-band cytochrome  $c_1$  is also presented.

## EXPERIMENTAL PROCEDURES

**Materials.** DEAE-cellulose (DE-52, Whatman), Ultrogel AcA 44 (LKB), Sephacryl S-200, Sephadex G-75 and G-25

(Pharmacia), and horse cytochrome  $c$  type III (Sigma) were purchased commercially. Sodium dodecyl sulfate (Fisher), cholic acid (Sigma), and deoxycholic acid (Sigma) were recrystallized in aqueous ethanol with the aid of active carbon. Acrylamide obtained from a commercial source was recrystallized in chloroform. Emasol 1130 was a gift from Kao Soap Co. (Tokyo). All other chemicals were commercially obtained in the highest grade available.

The Keilin-Hartree preparation (HMP)<sup>1</sup> of beef heart was obtained as previously reported (King, 1967). The succinate-cytochrome  $c$  reductase (Takemori & King, 1964) was prepared from the heart muscle preparation with the following modifications. The precipitate of crude succinate cytochrome  $c$  reductase was dissolved in 0.1 M phosphate buffer, pH 7.4, containing 0.5% cholate with a volume of one-sixth of the original HMP and then kept overnight (~16 h). The turbid red-brown solution was then centrifuged at 20 000 rpm for 1 h in a Beckman Model J-21B to remove the insoluble dark brown precipitate. The resulting clear red supernatant was brought to 33% ammonium sulfate saturation. Following 20-min incubation with stirring, the solution was centrifuged at 13 000 rpm for 30 min, and the supernatant was brought to 48% ammonium sulfate saturation. This was incubated and centrifuged as above, and the resulting precipitate was dissolved in 50 mM phosphate buffer, pH 7.4. The reductase thus obtained contained cytochrome  $c_1$  of 2.0 nmol/mg of protein or higher.

All the procedures for the purification described were performed at 0-4 °C unless otherwise indicated.

**Spectrophotometric Measurements.** All measurements were performed at room temperature (about 23 °C) except otherwise stated. Absorption spectra were recorded by using a Cary spectrophotometer, Model 14 or 16. The concentration of cytochrome  $c_1$  was determined spectrophotometrically, by

<sup>†</sup> This work was supported by National Institutes of Health Grant GM 16767 (to T.E.K.) and by a grant-in-aid from the American Heart Association (to C.H.K.).

\* Address correspondence to this author.

<sup>1</sup> Abbreviations: HMP, Keilin-Hartree heart muscle preparation; SDS-PAGE, sodium dodecyl sulfate-polyacrylamide gel electrophoresis.

using an extinction coefficient of  $17.5 \text{ mM}^{-1}$  for  $A_{\text{red}}^{552.5} - A_{\text{red}}^{540}$  (Yu et al., 1972). Absorption spectra at low temperature ( $-180^\circ\text{C}$ ) were recorded in a Cary Model 14 spectrophotometer equipped with a scattered transmission accessory, an RCA 06217 photomultiplier, and a Sylvania DWY quartzline lamp. The sample was placed in a cuvette of approximately 2-mm optical path and was cooled by liquid nitrogen. The temperature was directly recorded on a Leeds-Northrop Speedomax W strip-chart recorder with a calibrated copper-constantan thermocouple; one end of the thermocouple was inserted in the sample. The flow of nitrogen gas was constantly blown onto the surface of cuvette to avoid any ice buildup. The setting was checked with crystalline cytochrome *c* and gave the same result as reported (Estabrook, 1961). Circular dichroic spectra were recorded in a Cary spectropolarimeter 6000-6001 as previously described (Kim & King, 1981, 1983; Kaminsky et al., 1975) and expressed in degrees centimeter squared per decimole. Photoreduction was performed according to a previous report (Yu et al., 1975) using  $9.8 \mu\text{M}$  ferricytochrome *c*<sub>1</sub> in 50 mM phosphate buffer with a 750-W projector lamp. The reduction was measured by scanning the sample in the spectrophotometer at predetermined time intervals.

Polyacrylamide gel electrophoresis in sodium dodecyl sulfate was performed according to the methods of Weber and Osborn (1969) and Swank and Munkres (1971). In the former method, 10% gel plus 0.3% cross-linker and in the latter 12.5% and 0.44% were used, respectively. Sedimentation equilibrium was conducted in a Beckman Model E analytical ultracentrifuge. The experiment was made at  $20^\circ\text{C}$  with the scanner wavelength set at 417 nm. It was kindly performed by Dr. R. MacColl in the New York State Health Laboratory. Molecular weight was calculated as usual (Kim & King, 1983). Isoelectric focusing was carried out in a sucrose density gradient using 1% ampholine, pH range 3-6. The details of the method were described before (Kim & King, 1983), and redox potential measurements were conducted in a conventional manner using ascorbate as a reductant and Wurster's blue (Mustafa et al., 1968) as a mediator between cytochrome *c*<sub>1</sub> and the platinum electrode (Chiang, 1976);  $7 \mu\text{M}$  one-band cytochrome *c*<sub>1</sub> in 5 mL of 50 mM phosphate buffer, pH 7.4, was titrated with 1 mM sodium ascorbate at  $23^\circ\text{C}$ . Wurster's blue was present at  $20 \mu\text{M}$ , and reduction was measured at 417-434 nm. Measurement was performed in a closed vessel under a continuous flow of argon.

**Enzymatic Activity.** Enzymatic oxidation and reduction of cytochrome *c*<sub>1</sub>, i.e., as electron donor or acceptor, were estimated as previously reported (Yu et al., 1972); oxidation was conducted in 1 mL of reaction mixture of  $16.6 \mu\text{g}$  of the heart muscle preparation,  $0.75 \mu\text{M}$  cytochrome *c*, and the various concentrations of reduced one-band *c*<sub>1</sub> in 0.1 M phosphate buffer, pH 7.4. Reduction was conducted in 1 mL of reaction mixture of 20 mM succinate, 1.5 mM cyanide,  $0.75 \mu\text{M}$  cytochrome *c*,  $17 \mu\text{g}$  of the heart muscle preparation, and the various concentrations of oxidized one-band *c*<sub>1</sub>. Reaction was initiated by addition of the heart muscle preparation or succinate in the case of reduction.

## RESULTS

### Purification

The method reported here was based on the method by König et al. (1980) with modifications. This method was improved since our previous brief report (Kim & King, 1981) so that pure one-band cytochrome *c*<sub>1</sub> completely free of the hinge protein could be prepared with a reasonable yield.

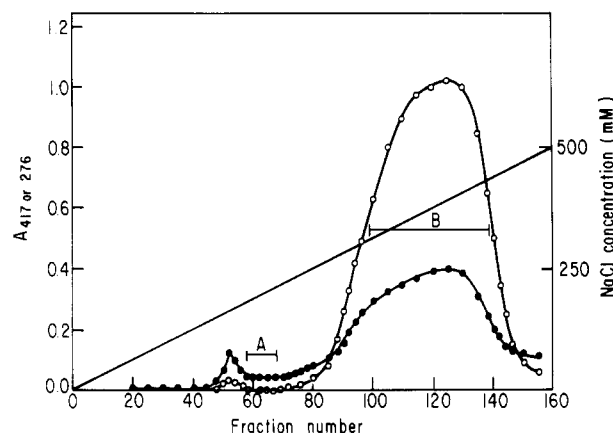


FIGURE 1: DEAE-cellulose column chromatography. The repeatedly washed *c*<sub>1</sub>-adsorbed DEAE-cellulose (80 mL) was packed on a cushion of DEAE-cellulose column ( $2.6 \times 10 \text{ cm}$ ) equilibrated with 50 mM potassium phosphate buffer containing 1% cholate and 0.1%  $\beta$ -mercaptoethanol. It was further washed with 2 bed volumes of the buffer. The linear gradient of 0.0-0.5 M NaCl in the same buffer was used to elute the cytochrome *c*<sub>1</sub>. The total volume of gradient buffer was 3.2 times the bed volume. The flow rate was 12.6 mL/h with each fraction of 4.2 mL. The closed circles represent  $A_{276}$ ; open circles indicate  $A_{417}$ . The absorbance was recorded in a cell with a 0.2-cm light path. "A" was mainly the hinge protein with the impurities, and "B" was used for subsequent Ultrogel AcA 44 column chromatography.

Succinate-cytochrome *c* reductase with a purity above 2.0 nmol of *c*<sub>1</sub>/mg, 3.7 g, was adjusted to a protein concentration of 11-13 mg/mL in 50 mM potassium phosphate buffer, pH 7.4, containing 1.5% potassium cholate, 0.5% potassium deoxycholate, 8% of ammonium sulfate saturation, and 15%  $\beta$ -mercaptoethanol with the addition of appropriate stock solutions. This was incubated for 1 h with slow stirring at  $4^\circ\text{C}$ . The extracted cytochrome *c*<sub>1</sub> from the reductase was separated from the cytochrome *b* by centrifugation at 20 000 rpm in a Beckman Model J-21B for 20 min. The supernatant was quickly passed through a Sephadex G-25 ("medium" size particle) column ( $5 \times 60 \text{ cm}$ ) which had been equilibrated with 50 mM potassium phosphate buffer, pH 7.4, mainly for desalting. The slightly turbid cytochrome *c*<sub>1</sub> fraction at the void volume was centrifuged as before for 20 min. The clear supernatant was immediately adjusted to 1% potassium cholate and adsorbed, in a beaker, to 80 mL of DEAE-cellulose equilibrated with 50 mM potassium phosphate buffer, pH 7.4, containing 1% potassium cholate and 0.1%  $\beta$ -mercaptoethanol. The *c*<sub>1</sub>-adsorbed DEAE-cellulose was washed on a 9-cm diameter sintered glass funnel, but not sucked to dryness, with the above buffer until the washing showed no absorbance at 280 nm. More than 2 L of buffer was usually required. The washed *c*<sub>1</sub>-adsorbed DEAE-cellulose was packed on the top of a DEAE-cellulose column ( $2.6 \times 10 \text{ cm}$ ) as a cushion equilibrated with the same buffer. The whole column ( $2.6 \times 25 \text{ cm}$ ) was then washed again with 2 bed volumes of the buffer. The elution was performed with a linear gradient of 0.0-0.5 M NaCl in the 50 mM potassium phosphate buffer, pH 7.4, containing 1% potassium cholate and 0.1%  $\beta$ -mercaptoethanol.

Fractions containing cytochrome *c*<sub>1</sub> (labeled "B" in Figure 1) were pooled and concentrated on an Amicon ultrafiltration cell, using a PM 30 Diaflo ultrafiltration membrane, to a protein concentration of 14-16 mg/mL. The concentrated *c*<sub>1</sub> fraction was applied on an Ultrogel AcA 44 column of  $2.6 \times 44 \text{ cm}$ , equilibrated with 50 mM potassium phosphate buffer, pH 7.4, containing 1% potassium cholate and 0.1%  $\beta$ -mercaptoethanol. The elution profile is shown in Figure 2, and

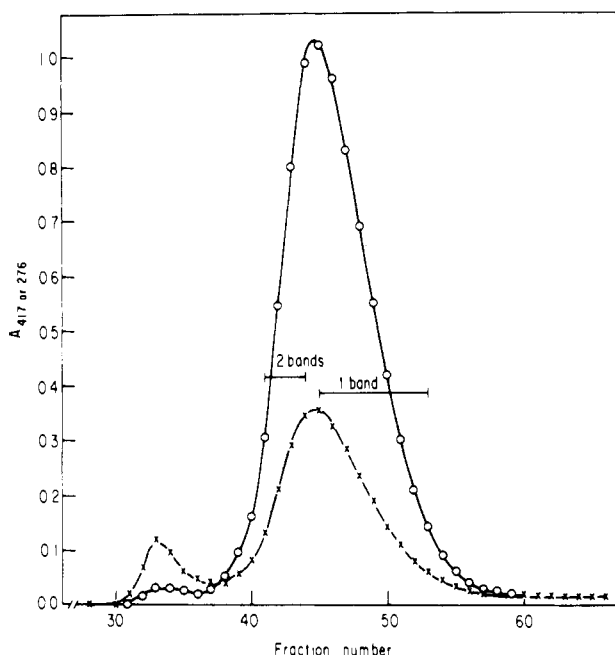


FIGURE 2: Ultrogel AcA 44 column chromatography. The cytochrome  $c_1$  preparation obtained from DEAE-cellulose chromatography was concentrated to 16 mg/mL and then applied to an Ultrogel AcA 44 column ( $2.6 \times 44$  cm). The elution was performed with 50 mM potassium phosphate buffer, pH 7.4, containing 1% cholate and 0.1%  $\beta$ -mercaptoethanol. The flow rate was 6 mL/h with each fraction of 2 mL. (X) Absorbance at 276 nm; (O) absorbance at 417 nm. All absorbance was recorded in a cell with a 0.2-cm light path. The first peak was impure cytochrome  $c_1$  with high molecular weight species as revealed by SDS-PAGE. The fractions labeled "2 bands" and "1 band" are purified cytochrome  $c_1$  preparations.

Table I: Purity and Yield of One-Band Cytochrome  $c_1$  from Succinate-Cytochrome  $c$  Reductase by the Improved Method

purification step	$A_{417}/A_{276}$	purity (nmol/mg)	cytochrome $c_1$ ( $\mu$ mol)	yield (%)
succinate-cytochrome $c$ reductase		2.0-2.2	7.4	100
Sephadex G-25 column	1.1		4.8	65
DEAE-cellulose batchwise washing	2.5		3.4	45
Ultrogel AcA 44 chromatography				
two-band $c_1$	2.4-2.7	25-27	1.1	15
one-band $c_1$	2.9-3.0	31-32	1.9	25
total $c_1$				40

the purification steps are summarized in Table I. It was found that the yield of one-band cytochrome  $c_1$  was increased by further washing of  $c_1$ -adsorbed DEAE-cellulose and the yield of two-band  $c_1$  was decreased concomitantly. This fact suggested that the heme-containing polypeptide and the hinge protein of two-band  $c_1$  were associated with each other around neutral pH and the latter could be removed by washing. A decrease in the washing step of  $c_1$ -adsorbed DEAE-cellulose increased the yield of two-band cytochrome  $c_1$ . In other words, the hinge protein which was tightly associated with the cytochrome  $c_1$  was very slowly washed away from the  $c_1$  fraction on a DEAE-cellulose column. However, sufficient washing on the sintered glass funnel was required in order to remove other impurities as much as possible and to obtain even pure two-band  $c_1$ . Nevertheless, the yield of one-band  $c_1$ , at a purity of 32 nmol/mg of protein, was usually not more than 25%, and that of two-band  $c_1$ , with a purity of 25-27 nmol of heme/mg of protein, could reach as high as 40%. In general,

the total yield of these two species of  $c_1$  was found always to be approximately 40%. It was observed that if the sample for Ultrogel AcA 44 chromatography were concentrated too high ( $>18$  mg/mL), the yield of one-band  $c_1$  decreased, probably because some aggregation took place.

It may be mentioned that there are two methods for the preparation of two-band  $c_1$ . The one reported in 1972 [Yu et al., 1972; also cf. King (1978)] is referred to as the original method. The other is the one described in this paper.

### Properties

**General Properties of Purified Cytochrome  $c_1$ .** The purified one-band  $c_1$  thus obtained was found completely free of the hinge protein. Two-band  $c_1$  which showed a purity higher than 25 nmol of heme/mg of protein consisted of only two band on SDS-PAGE (Kim & King, 1983); one was  $M_r$  29 000, and the other was  $M_r$  11 000. The amino acid composition of one-band  $c_1$  was in agreement with that from the sequence analysis (Wakabayashi et al., 1980, 1982b). Both two-band  $c_1$  and one-band  $c_1$  were found to be enzymically active. They did not react with carbon monoxide at all. However, one-band  $c_1$  did not form the  $c_1$ - $c$  complex with cytochrome  $c$  (Kim & King, 1981, 1983), while two-band  $c_1$  did form the complex.

**Molecular Weight.** An analysis of SDS-PAGE revealed the molecular weights of one-band  $c_1$  and the heme peptide of two-band  $c_1$  to be 29 000. From the sedimentation equilibrium experiments, the molecular weight of one-band cytochrome  $c_1$  was found to be 44 000 when the sample at 2  $\mu$ M in 50 mM potassium phosphate buffer, pH 7.4, containing 1% cholate was run at 30 000 rpm for 20 h. The molecular weight obtained was higher than that obtained by the sequence analysis (Wakabayashi et al., 1980, 1982), which was apparently due to micelle and/or complex formation with detergent. Gel filtration chromatography on Sephacryl S-200 or Ultrogel AcA 44 of one-band and two-band  $c_1$  in 50 mM phosphate buffer, pH 7.4, without detergent showed a molecular mass of  $\sim 240$  000 for both one-band and two-band  $c_1$ , indicating one-band  $c_1$  to be an octamer and two-band  $c_1$  to be a hexamer in the absence of detergent. These highly polymerized cytochromes  $c_1$ , however, were easily depolymerized after dialysis in the presence of detergent. In contrast, the two-band  $c_1$  prepared by the original method was not depolymerized by detergent [cf. King (1983)].

**Isoelectric Point.** Isoelectric focusing of two-band  $c_1$  was performed in a sucrose density gradient using a pH gradient range of 3-6 and showed the  $pI$  value for the hinge protein to be 3.9 as reported (Kim & King, 1983) and the  $pI$  for  $c_1$  (heme peptide) to be 5.8, although the  $pI$  value for  $c_1$  by this method is approximate because of its precipitation at such low pH. Nevertheless, the observed value seemed to be close enough to the actual value, since even after 6 h at 500 V, the position of the red sharp band was not changed. It may be mentioned that the calculated value of  $pI$  should be higher since it contained 36 basic amino acids (12 Lys, 9 His, and 15 Arg) and only 29 acidic amino acids. The discrepancy is evidently due to the formation of  $c_1$ -detergent or  $c_1$ -phosphate-detergent complexes in which perhaps the basic residues are buried in the interior.

**Absorption Spectra.** Both preparations of two-band  $c_1$  and one-band  $c_1$  in the oxidized and reduced forms showed the same absorption spectra (cf. Figure 3A). One-band cytochrome  $c_1$  in the oxidized form also showed a broad peak at 690 nm as that in two-band  $c_1$ , and this broad band was abolished after reduction.

At  $-180^\circ\text{C}$ , the reduced cytochrome  $c_1$  exhibited a colorful spectrum as shown in Figure 3B-D, and essentially (see

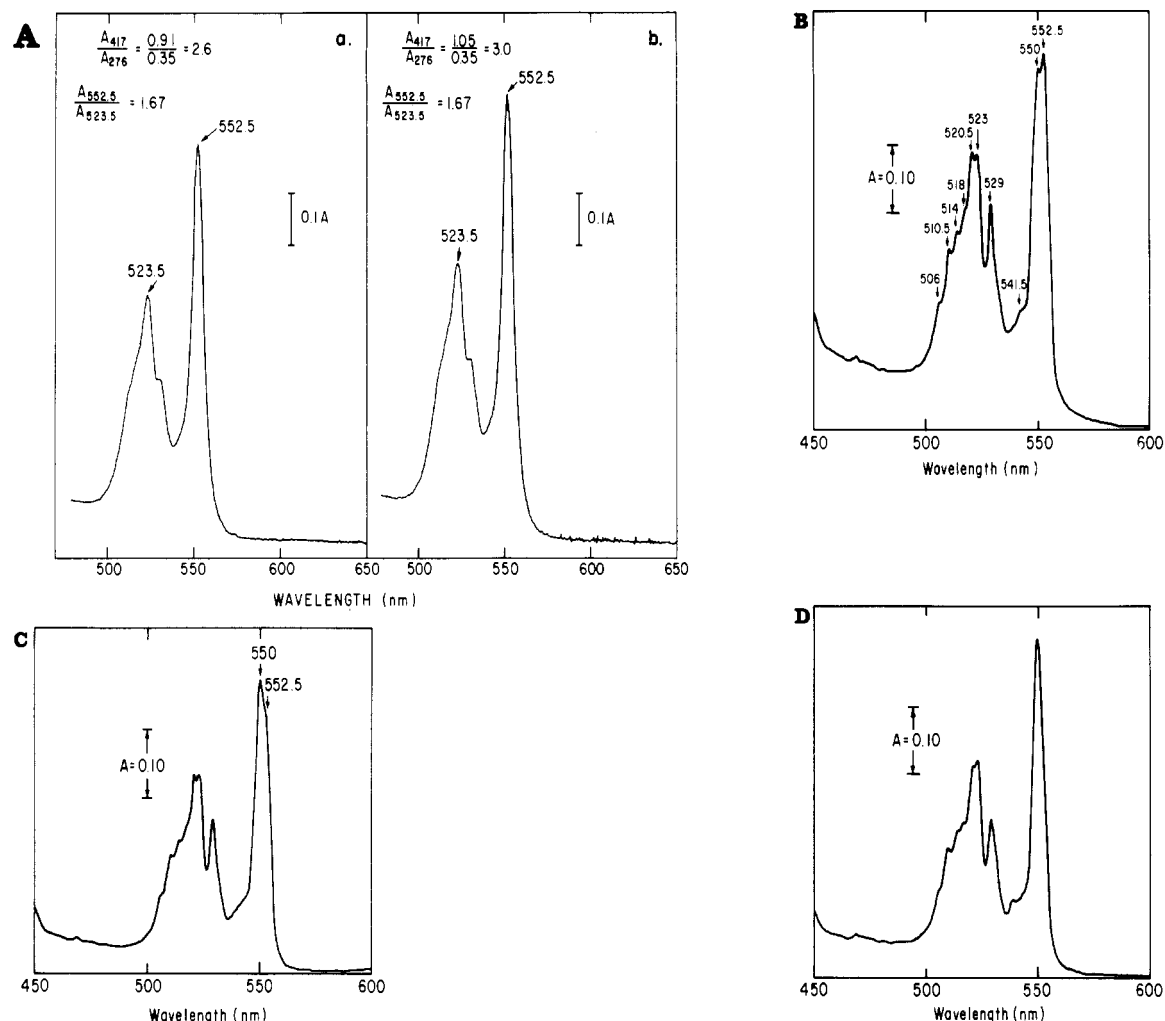


FIGURE 3: (A) Absorption spectra of one-band and two-band cytochrome  $c_1$ . (a) Two-band  $c_1$  and (b) one-band  $c_1$  in 50 mM potassium phosphate buffer, pH 7.4, containing 1% cholate and 0.1%  $\beta$ -mercaptoethanol. (B–D) Low-temperature spectra of one-band cytochrome  $c_1$ . (B) 61  $\mu$ M in 50 mM phosphate buffer, pH 7.4; (C) 60  $\mu$ M in 50 mM phosphate buffer, pH 7.4, containing 1% cholate and 0.1%  $\beta$ -mercaptoethanol; (D) 60  $\mu$ M in 50 mM phosphate buffer, pH 7.4, containing 0.5% Tween 80. The sample was cooled in a gold-plated UV-transparent polyethylene cuvette in liquid nitrogen. The approximate cuvette temperature was  $-180^\circ\text{C}$ ; the optical path was about 2 mm. Details of experiment were described under Experimental Procedures.

Discussion) no difference was observed between one-band  $c_1$  and two-band  $c_1$ . The  $\alpha$ -band, which showed only one band of 552.5 nm at room temperature, split into two absorption maxima at 552.5 and 550 nm with a distinct shoulder at 541.5 nm. The  $\beta$ -band, which showed three bands of absorption maxima at room temperature, displayed a spectrum showing maxima at 529, 523, 520.5, 518, 514, 510.5, and 506 nm.

However, the low-temperature splitting of both  $\alpha$  and  $\beta$  maxima was dependent on the presence of detergent. When cytochrome  $c_1$  was in the buffer medium containing 1% cholate and 0.1%  $\beta$ -mercaptoethanol, the  $\alpha$ -band showed some change from that (Figure 3B) of  $c_1$  in just buffer without detergent; i.e., the 552.5-nm peak almost diminished with a complete disappearance of the 541.5-nm band as presented in Figure 3C. When cholate and  $\beta$ -mercaptoethanol were removed from the cytochrome  $c_1$  sample by dialysis, spectrum B (Figure 3) was recovered again. Figure 3D shows a spectrum obtained when cytochrome  $c_1$  was in the buffer containing 0.5% Tween 80. The presence of 0.5% Tween 80 completely abolished the 552.5-nm peak from the  $\alpha$ -band and the 518-nm peak from the  $\beta$ -band, and a new 538-nm peak in the  $\alpha$ -band region and a 516-nm peak in the  $\beta$ -band region appeared. This detergent dependence on the low-temperature spectrum of cytochrome  $c_1$  was observed in both one-band and two-band  $c_1$  prepared by the present method. The low-temperature spectrum from

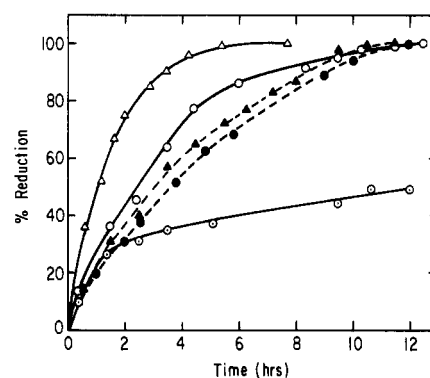


FIGURE 4: Photoreduction of cytochrome  $c_1$ . Cytochromes were photoreduced under anaerobic conditions as described under Experimental Procedures. The concentration of cytochrome  $c_1$  is 9.8  $\mu$ M, and that of cytochrome  $c$  is 10  $\mu$ M. (Δ) One-band  $c_1$  and (O) two-band  $c_1$  in 50 mM phosphate buffer, pH 7.4; (▲) one-band  $c_1$  and (●) two-band  $c_1$  in 50 mM phosphate buffer, pH 7.4, containing 1% Emasol 1130; (○) cytochrome  $c$  in 50 mM phosphate buffer, pH 7.4.

two-band cytochrome  $c_1$  prepared by the original method was not affected by detergent (King, 1983).

**Photoreduction.** Photoreductions of one-band  $c_1$  and two-band  $c_1$  were performed at pH 7.4 as shown in Figure 4. The

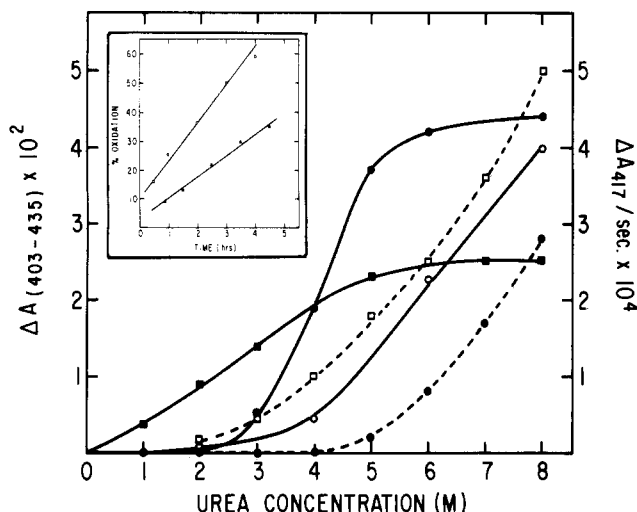


FIGURE 5: Effect of urea concentration on the Soret absorption and autooxidation of cytochromes  $c_1$  and  $c$ . (■) 1.2  $\mu$ M one-band ferri-cytochrome  $c_1$ , (●) 1.2  $\mu$ M two-band ferri-cytochrome  $c_1$ , and (○) 1.3  $\mu$ M ferri-cytochrome  $c$  for the Soret absorbance on the left ordinate; (□) 2.0  $\mu$ M one-band ferrocytochrome  $c_1$  and (●---●) 2.1  $\mu$ M two-band ferrocytochrome  $c_1$  on the right ordinate. All systems were in 0.1 M phosphate buffer, pH 7.4, 23 °C. Inset: Comparison of autooxidation rate of one-band  $c_1$  and two-band  $c_1$ . (○) 6.3  $\mu$ M one-band  $c_1$ ; (Δ) 6.4  $\mu$ M two-band  $c_1$ . All systems are in 50 mM phosphate buffer, pH 7.4, 23 °C.

rate of photoreduction in the absence of detergent (Emasol 1130) was much greater in one-band  $c_1$  than in two-band  $c_1$ . We observed complete reduction of one-band  $c_1$  at 6 h while two-band  $c_1$  required 10 h. However, in the presence of Emasol 1130, the rate of photoreduction was decreased, showing complete reduction of both one-band and two-band  $c_1$  at approximately 11 h, with no distinctive differences between them. It was obvious that the effect of Emasol 1130 on two-band  $c_1$  was much smaller than that on one-band  $c_1$ . As a control, photoreduction of cytochrome  $c$  was conducted, and the results (cf. Figure 4) showed maximal reduction of only about 50% in 12 h. These results were in general agreement with those reported for two-band  $c_1$  prepared by the original method (Yu et al., 1975).

**Effect of Urea Concentration on Cytochrome  $c_1$ .** As depicted in Figure 5, the denaturation of the oxidized cytochromes  $c_1$  was monitored at the various urea concentrations shown. The rate of autooxidation of ferrocytochromes  $c_1$  was also measured at the various urea concentrations. In the oxidized form of two-band  $c_1$ , an absorbance change of  $\Delta A_{403-435}$  occurred at 3 M urea, and this change leveled off at 6 M urea. However, the oxidized form of one-band  $c_1$  was more unstable than that of two-band  $c_1$ , showing a gradual change even at the lower urea concentration, but the extent of change was smaller than that of two-band  $c_1$  at higher urea concentrations. In the case of ferrocytochromes  $c_1$ , the effect of urea occurred at 5 M urea in two-band  $c_1$  and at 2 M urea in one-band  $c_1$ , where one-band  $c_1$  was again more sensitive than two-band  $c_1$ .

The stability of one-band and two-band cytochrome  $c_1$  was further examined by observing the rate of oxidation of the reduced forms of one-band and two-band  $c_1$  at room temperature. The inset of Figure 5 shows the extent of oxidation of reduced one-band and two-band  $c_1$  at the given time. Autooxidation of one-band  $c_1$  was about 2 times faster than that of two-band  $c_1$ . These differences between one-band and two-band  $c_1$  in urea treatment and autooxidation rate suggested that the conformations of these species might not be the same. The CD results substantiated this hypothesis. Circular dichroic

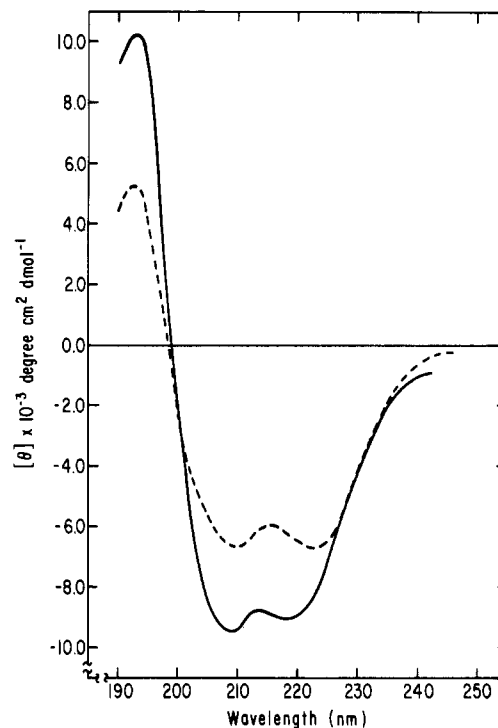


FIGURE 6: Circular dichroic spectra of the cytochrome  $c_1$  in the UV region at pH 7.4 in 50 mM phosphate buffer containing 1% Emasol 1130: (---) one-band cytochrome  $c_1$ ; (—) "two-band" cytochrome  $c_1$ .

spectra in the UV region (Figure 6) showed a significant difference between these two cytochromes  $c_1$  [cf. King et al. (1983)], although those in the Soret region exhibited the same behavior.

**Interaction with Cyanide.** Investigation of a possible reaction of cyanide with cytochrome  $c_1$  at pH 7.4 was conducted, and it was found that there was no reaction either with one-band or with two-band  $c_1$ . However, after cytochrome  $c_1$  was denatured, for example, by sodium dodecyl sulfate, a reaction of  $c_1$  with cyanide occurred. The denaturation was effected with as low as 0.03% sodium dodecyl sulfate. However, when cytochrome  $c_1$  was denatured with 0.3% or higher sodium dodecyl sulfate, even dithionite could not reduce this cytochrome  $c_1$ . Figure 7 shows spectra of the 0.1% SDS-denatured one-band cytochrome  $c_1$  in oxidized and dithionite-reduced forms, and its spectra after cyanide reaction are also depicted. Red shifts, ranging from 3 to 5 nm, in the  $\alpha$ ,  $\beta$ , and Soret regions were observed in the ferrocytochrome  $c_1$ -cyanide complex. These results are the same as those of the previous report of two-band  $c_1$  prepared by the original method (Kaminsky et al., 1975). It should be noted that the present preparations of one-band and two-band cytochrome  $c_1$  are more sensitive to SDS than two-band  $c_1$  prepared by the original method.

**Redox Potential of One-Band Cytochrome  $c_1$ .** The midpoint potential ( $E_m$  at 7.4) for one-band  $c_1$  was found to be +227 mV and was close to the value reported for two-band  $c_1$  prepared by the original method (Chiang, 1976) and that of  $c_1$  in submitochondrial particles (Dutton et al., 1970).

**Enzymatic Oxidation and Reduction of Cytochrome  $c_1$ .** Both the enzymatic oxidation and reduction of one-band cytochrome  $c_1$  were determined with the heart muscle preparation as the electron carrier. Cytochrome  $c_1$  in the reduced form was used as an electron donor and oxygen as an acceptor in oxidation, and in reduction, cytochrome  $c_1$  in the oxidized form was used as an acceptor and succinate as a donor in the presence of a terminal inhibitor.

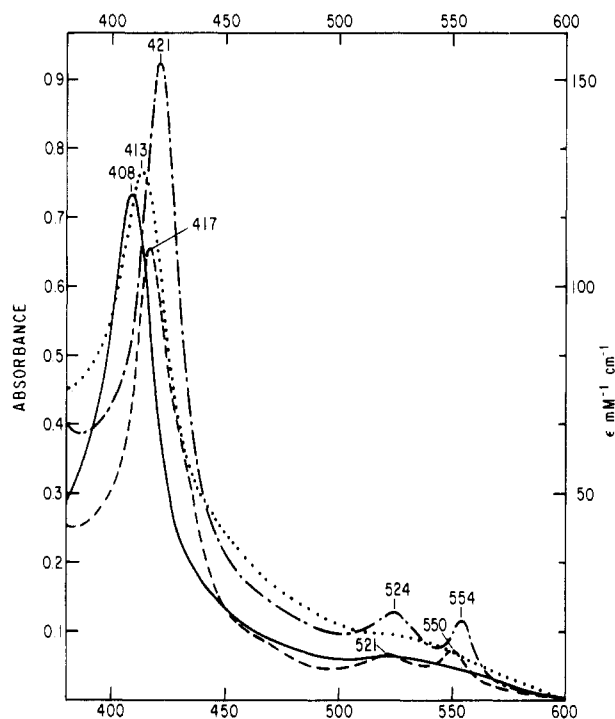


FIGURE 7: Spectra of cyanide-reacted cytochrome  $c_1$  after denaturation. One-band cytochrome  $c_1$  ( $7 \mu\text{M}$ ) was denatured by 0.1% sodium dodecyl sulfate in 50 mM phosphate buffer, pH 7.4: (—) oxidized form; (---) reduced form; (...) oxidized form after cyanide reaction; (-.-) reduced form after cyanide reaction. Freshly prepared cyanide solution (neutralized) was added to make 0.3 M after SDS denaturation.

The oxidation of the reduced cytochrome  $c_1$  by oxygen in the presence of the heart muscle preparation was dependent on the concentration of cytochrome  $c_1$ . Under these conditions, the  $K_m$  for one-band cytochrome  $c_1$  was found to be  $1.33 \mu\text{M}$  and the  $V_{\max}$  to be 273 nmol of cytochrome  $c_1$  per minute per milligram of the heart muscle preparation at  $23^\circ\text{C}$ . The reduction of oxidized cytochrome  $c_1$  by succinate catalyzed by the heart muscle preparation was also dependent on the concentration of  $c_1$ , showing the  $K_m$  for cytochrome  $c_1$  to be  $4.54 \mu\text{M}$  with a  $V_{\max}$  of 200 nmol of cytochrome  $c_1$  reduced per minute per milligram of the heart muscle preparation. These results indicated the cytochrome  $c_1$  samples were enzymatically active.

#### DISCUSSION

It should be emphasized that the yield of one-band  $c_1$  increased by more washings on DEAE-cellulose with the sacrifice of that of two-band  $c_1$ . Nevertheless, insufficient washing may impair the purity of both two-band and one-band  $c_1$  due to the incomplete removal of impurities. The protein concentration of concentrated  $c_1$  fraction before the Ultrogel AcA 44 column seemed to be important for both the purity and yield of one-band and two-band cytochromes  $c_1$ . A protein concentration of 14–16 mg/mL usually gave good results and should not be more than 18 mg/mL.

The molecular weight of one-band cytochrome  $c_1$  or a heme-containing subunit of two-band  $c_1$  was estimated to be 29 000 by SDS-PAGE, and this is very close to  $M_r$  27 924 obtained by the direct amino acid sequence determination (Wakabayashi et al., 1980, 1982b). The sedimentation equilibrium technique shows one-band cytochrome  $c_1$  to be  $M_r$  44 000 in 50 mM potassium phosphate buffer, pH 7.4, containing 1% cholate. König et al. (1980) have also reported  $M_r$  45 000 by the ultracentrifuge technique and  $M_r$  31 000 by SDS-PAGE. The higher value is possibly due to the formation

of a cytochrome  $c_1$ -cholate complex and probably also a  $c_1$ -phosphate-detergent complex since phosphate binds with cytochrome  $c$  [see Taborsky & McCollum (1979) and footnote and references cited therein]. This hypothesis is consistent with the disparity between the observed and calculated isoelectric points. The recent report by Shimomura et al. (1985) showed  $M_r$  25 600 calculated from the specific heme content and  $M_r$  33 000 by the SDS-PAGE technique of Merle and Kadenbach (1980). Although the results of the different SDS-PAGE techniques allow the difference in the estimation of molecular weight, the value of  $M_r$  25 600 from the specific heme content is rather low compared to the sequence data (Wakabayashi et al., 1980, 1982b). This could be one of the reasons why the purity of their cytochrome  $c_1$  preparation was found to be 39 nmol/mg of protein even though the ratio of  $A_{417}/A_{277}$  was 2.31 which was also smaller than our result (see Table I).

Absorption spectra of one-band  $c_1$  and two-band  $c_1$  at either room temperature or liquid nitrogen temperature (Figure 3) do not show significant differences, implying the association of the hinge protein to cytochrome  $c_1$  does not greatly change the electronic structure of cytochrome  $c_1$  (Kim & King, 1981, 1983). However, one interesting observation from the spectra at liquid nitrogen temperature is that the  $\alpha$ -band shows a split 552.5- and 550-nm peak (Figure 3B) without an additional 549-nm peak which is observed on the two-band  $c_1$  prepared by the original method (Yu et al., 1972), and these bands are altered by detergents; with cytochrome  $c_1$  associated with 1% cholate in 50 mM phosphate buffer, pH 7.4, the 552.5-nm peak was diminished, but the 550-nm peak was enhanced (Figure 3C), and this was demonstrated reversible by the removal of cholate. Moreover, 0.5% Tween 80 can completely abolish the peak at 552.5 nm (Figure 3D). We are inclined to consider this observation to be related to its polymerized state; probably completely polymerized cytochrome  $c_1$  shows two distinct peaks of 550 and 552.5 nm, and the presence of 1% cholate can depolymerize cytochrome  $c_1$ , even though not completely, resulting in a decrease of the 552.5-nm peak. The 550-nm peak in the presence of Tween 80 may support our interpretation considering that cytochrome  $c_1$  can be completely monomerized in Tween 80. This is further substantiated by the fact that two-band  $c_1$  prepared by the original method (Yu et al., 1972; King, 1978) shows practically no effect of the detergent on  $\alpha$ -band splitting. The disparity may be due to the more tightly bound polymerized form in these samples prepared by the original method [see references cited in King (1983)].

Our present cytochrome  $c_1$  preparation exhibited similar photoreduction behavior as that shown in cytochrome  $c_1$  prepared by the original method (Yu et al., 1975). The profile shown in Figure 4 may reflect some difference in the reactivity of one-band and two-band cytochrome  $c_1$ . Obviously, one-band  $c_1$  is more sensitive to the light than two-band  $c_1$  probably because the heme of one-band  $c_1$  is not as fully covered by the protein moiety as that of two-band  $c_1$ . This is further demonstrated by the effect of the presence of Emasol 1130.

The reaction of cyanide with cytochrome  $c_1$  is observed only when cytochrome  $c_1$  is denatured (Kaminsky et al., 1975) while the reaction between the native form of cytochrome  $c$  and cyanide does take place (George & Tsou, 1952). Here again, the observation differed from the report by Schejter and Berke (1968).

The midpoint potential (+227 mV) for one-band  $c_1$  is practically the same as that of two-band  $c_1$  (+225 mV) (Chiang, 1976), and this suggests that the hinge protein does

not affect the redox potential ( $E_m$ ) of cytochrome  $c_1$ . This observation is in agreement with the fact that no direct participation of the hinge protein is observed in the electron-transfer reaction between cytochromes  $c_1$  and  $c$  (Kim et al., 1984) in rapid kinetic studies. Moreover, the enzymatic activity of present cytochrome  $c_1$  preparations shows that either one-band or two-band cytochrome  $c_1$  is active to serve as an electron donor as well as acceptor.

In conclusion, we have developed a simple and rapid method for the isolation of highly purified one-band and two-band cytochrome  $c_1$  with substantially improved yield. The two-band cytochrome  $c_1$  thus obtained is found to be a convenient starting material for the preparation of the hinge protein. Differences between two-band and one-band  $c_1$  in certain properties may suggest a probable role of the hinge protein for stabilizing the conformation of cytochrome  $c_1$  in the mitochondrial respiratory chain.

#### ACKNOWLEDGMENTS

We acknowledge the valuable assistance of M. Seaman and S. Chace. We are most grateful to Dr. R. MacColl for the generous and careful study of the sedimentation equilibrium of one-band cytochrome  $c_1$ .

**Registry No.**  $\text{NH}_2\text{CONH}_2$ , 57-13-6;  $\text{CN}^-$ , 57-12-5; cytochrome  $c_1$ , 9035-42-1.

#### REFERENCES

- Chiang, Y. L. (1976) Ph.D. Thesis, State University of New York at Albany.
- Dutton, P. L., Wilson, D. F., & Lee, C. P. (1970) *Biochemistry* 9, 5077-5082.
- Estabrook, R. W. (1961) in *Hematin Enzymes* (Falk, J. E., Lemberg, R., & Morton, R. K., Eds.) pp 436-460, Pergamon Press, London.
- George, P., & Tsou, C. L. (1952) *Biochem. J.* 50, 440-448.
- Kaminsky, L. S., Chiang, Y. L., & King, T. E. (1975) *J. Biol. Chem.* 250, 7280-7287.
- Keilin, D. (1966) *History of Cell Respiration and Cytochrome*, Cambridge University Press, Cambridge.
- Kim, C. H., & King, T. E. (1981) *Biochem. Biophys. Res. Commun.* 101, 607-614.
- Kim, C. H., & King, T. E. (1983) *J. Biol. Chem.* 258, 13543-13551.
- Kim, C. H., Balny, C., & King, T. E. (1984) *Proc. Natl. Acad. Sci. U.S.A.* 81, 2026-2029.
- King, T. E. (1967) *Methods Enzymol.* 10, 202-208.
- King, T. E. (1978) *Methods Enzymol.* 53, 181-191.
- King, T. E. (1983) *Adv. Enzymol. Relat. Areas Mol. Biol.* 54, 267-366.
- King, T. E., Wu, C.-S. C., Yang, J. T., Matsubara, M., & Kim, C. H. (1983) *Fed. Proc., Fed. Am. Soc. Exp. Biol.* 42, 2070.
- König, B. W., Schilder, L. T. M., Tervoort, M. J., & van Gelder, B. F. (1980) *Biochim. Biophys. Acta* 621, 283-295.
- Merle, P., & Kadenbach, B. (1980) *Eur. J. Biochem.* 105, 499-507.
- Mustafa, M. G., Cowger, M. L., Labbe, R. F., & King, T. E. (1968) *J. Biol. Chem.* 243, 1908-1918.
- Schejter, A., & Berke, G. (1968) *Biochim. Biophys. Acta* 162, 459-461.
- Shimomura, Y., Nishikimi, M., & Ozawa, T. (1985) *J. Biol. Chem.* 260, 15075-15080.
- Swank, R. T., & Munkres, K. D. (1971) *Anal. Biochem.* 39, 462-477.
- Taborsky, G., & McCollum, K. (1979) *J. Biol. Chem.* 254, 7069-7075.
- Takemori, S., & King, T. E. (1964) *J. Biol. Chem.* 239, 3546-3558.
- Wakabayashi, S., Matsubara, H., Kim, C. H., Kawai, K., & King, T. E. (1980) *Biochem. Biophys. Res. Commun.* 97, 1548-1554.
- Wakabayashi, S., Takeda, H., Matsubara, H., Kim, C. H., & King, T. E. (1982a) *J. Biochem. (Tokyo)* 91, 2077-2085.
- Wakabayashi, S., Matsubara, H., Kim, C. H., & King, T. E. (1982b) *J. Biol. Chem.* 257, 9335-9344.
- Weber, K., & Osborn, M. (1969) *J. Biol. Chem.* 244, 4406-4412.
- Yakushiji, E., & Okunuki, K. (1940) *Proc. Imp. Acad. (Tokyo)* 16, 299-302.
- Yu, C. A., Yu, L., & King, T. E. (1972) *J. Biol. Chem.* 247, 1012-1019.
- Yu, C. A., Chiang, Y. L., Yu, L., & King, T. E. (1975) *J. Biol. Chem.* 250, 6218-6221.